30. ISOTOPIC GEOCHEMISTRY OF LAVAS FROM SITES 553 AND 555

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and
P. J. Hamilton, Scottish Universities Research and Reactor Centre

ABSTRACT

It is demonstrated by K-Ar analyses that the age of reversely magnetized basalts, which immediately predate magnetic Anomaly 24B, is 53.5 ± 1.9 m.y. Samples from deep levels appear to be grossly contaminated by an extraneous argon component with a uniform argon-40/argon-36 ratio ≥ 440. This component is thought to have been derived from fluids circulating in the lava pile during burial. The age result corroborates the assignment previously made to Anomaly 24B by Hailwood et al. (1979) and Lowrie and Alvarez (1981). It additionally suggests that lava extrusion formed part of a much larger magmatic event, which affected wide areas of the North Atlantic margins around the Palaeocene/Eocene boundary, and can therefore probably be considered a good estimate of the age of this boundary. Initial 143Nd/144Nd ratios lie in the very restricted range 0.512920 ± 19 to 0.513026 ± 24 and initial 87Sr/86Sr ratios from ca. 0.703 to ca. 0.705. Acid leaching reduces the latter range to 0.70264 ± 4 to 0.70384 ± 4, suggesting that the higher Sr/Sr ratios resulted from interaction with seawater. The array of data for treated samples is closely conformable on a 143Nd/144Nd-87Sr/86Sr diagram with the main oceanic mantle array and with previously published fields for Atlantic Ocean basalts. No evidence for any continental crustal contamination has been found. This suggests, but does not prove, that continental crust played no part in the genesis of these rocks.

INTRODUCTION

K-Ar, Rb-Sr, and Sm-Nd isotopic data have been obtained for basaltic rocks returned from DSDP Leg 81 to assess their age, origin, and the nature of their source region compared with other basalts from the North Atlantic region and possible evidence for any interaction with continental crust. Such information is prerequisite to a proper understanding of the early rifting and onset of spreading. In general the basalts are submarine with many pillows and are interbedded with tuffs, agglomerates, and sediments. At Holes 552A and 554A, only thin basalt sequences were encountered. The samples analyzed in this study were recovered from Sites 553 and 555. They are the seven least-altered samples from over 100 samples which have been analyzed in thin section by Harrison et al. (this volume), namely, two basalts from Site 553 and three basalts and two dolerites from Site 555. For clarity these samples have been simply numbered A through G. The corresponding DSDP numbers are listed in Appendix A to this chapter.

At Site 553 the lowermost basalts penetrated below the Tertiary sediments correspond to the dipping reflectors, the outer high of which is at the edge of Anomaly 24B. At Site 555 the basalts which are underlain by (intrusive?) olivine dolerites may also represent these reflectors. The basalts are reversely magnetized. Petrographically and chemically they are hypersthene-normative tholeiites. Despite the relative freshness of the samples analyzed they have undoubtedly, as shown elsewhere (Harrison et al., this volume), interacted with seawater, and the deepest rocks at Site 555 have experienced greenschist facies alteration.

ANALYTICAL PROCEDURE

The samples supplied were crushed to small (~1 mm) fragments, and these were used for argon analyses. A portion of each was ground in a Tema mill and the powders used for the potassium determinations. Further aliquots of the chips were cleaned ultrasonically in distilled water, dried, and ground. The resulting powders were used for Rb, Sr, Sm, and Nd analyses.

Potassium analyses were performed in duplicate by flame photometry on a Corning-Eel Model 450 instrument. A lithium internal standard and sodium buffering were employed. The mean error of the duplicate analyses on the seven samples listed in Table 1 is 0.688% ± 0.308 (2σ). A much more conservative estimate of ±1.5% has however been assigned to calculate the errors in K-Ar age (see footnote, Table 1).

Argon-40 analyses were performed by fusion in vacuo and gas purification employing (mainly) a Ti-Mo getter. The argon was measured statically by isotope dilution on an MSIO mass-spectrometer. The spike (argon-38, Zurich, 99.99%) was obtained from a pipette delivering ~8 × 10^-7 scc and periodically calibrated against an interstate biotite (BS133, 4.13 × 10^-4 scc/g), itself originally calibrated against known volumes of air in this laboratory. The spike heights were corrected for fractionation through the inlet leak valve in a manner similar to that described by Baksi and Farrar (1973). The amplifier ranges were calibrated and peak heights measured manually from a strip chart recorder. Error assignments are listed in the footnote to Table 1.

Concentrations of Rb, Sr, Sm, and Nd and 87Sr/86Sr and 143Nd/144Nd ratios were determined for single dissolutions of 50 mg aliquots of sample powder. In addition separate sample aliquots were leached with 6 M hydrochloric acid and 87Sr/86Sr ratios determined for the residues and leachates. The Rb, Sr, and Sm together with Nd were separated by conventional cation-exchange chromatography and the Sm and Nd further purified following techniques modified from Richard et al. (1976) and Zindler et al. (1979). Concentration determinations by isotope dilution were made on a Vacuum Generators Micromass 30 B mass spectrometer and isotope ratio measurements on a Vacuum Generators Isomass 54 E mass spectrometer.
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RESULTS

Potassium–Argon Dating

Since the samples have very low potassium contents and are (relatively) youthful, it was anticipated that some difficulty might be encountered in accurately measuring their small radiogenic argon-40 concentrations. Complications were also expected, under these conditions, the presence of any extraneous argon component of non-atmospheric isotopic composition, even in minute concentrations, would render the normal nonradiogenic correction invalid and grossly distort the conventionally calculated ages. A series of preliminary measurements was therefore carried out on small sample aliquots to investigate the possible extent of these problems. The results of these exploratory analyses are presented in Table 1.

It is immediately apparent that many of the conventionally calculated ages are characterized by large uncertainties (as a result of the high nonradiogenic argon-40 concentrations) and many are probably, in addition, anomalous. This seems particularly true for samples B, E, F, and G from the deeper levels at both sites. However the presence of abnormally high total concentrations of both argon-40 and argon-36 was also revealed in these analyses (cf. Fig. 2; circled points). This is interpreted as indicative of the presence of an extraneous argon component which has a non-atmospheric isotopic composition. No significance is therefore attached to the conventional ages calculated for samples B, E, F, and G. However the data from samples A, C and D in Table 1 can be used to give a first approximation to the age of the uppermost samples. When these are plotted in the manner suggested by York et al., 1969 (see also York and Farquhar, 1972, p. 30), a five-point isochron corresponding to an age of 59.0 ± 3.1 m.y. is obtained (see Appendix C).

To determine a more accurate value for the age of the samples and to attempt to investigate more fully the nature and origin of the extraneous argon, a further series of analyses were performed. In these the sample weights were significantly increased and no sample encapsulation was employed. Both measures were aimed at reducing the relative effect of atmospheric argon contamination in an attempt to surmount the problems encountered in the first series of analyses. These results are presented in Table 2.

Rubidium–Strontium and Samarium–Neodymium Results

Table 3 lists the results of these measurements. The abundances of Rb, Sr, Sm, and Nd are comparable with those of recent Atlantic MORB (e.g., Cohen et al., 1980; Cohen and O’Nions, 1982). The 147Sm/144Nd ratios are significantly increased and no sample encapsulation of these elements and the isotopic compositions of Sr and Nd.

Petrogenesis of the Basalts

The measured 87Sr/86Sr and 143Nd/144Nd ratios in Table 3 have been corrected for radioactive decay since 53 m.y. ago (see below), although for Sr the corrections are so small that they are within the limits of error for the measured isotopic ratio. The initial 143Nd/144Nd ratios are plotted against initial 87Sr/86Sr ratios on Figure 1 where they can be compared with the data fields (estimated for 53 m.y. ago) for unaltered Atlantic MORB and for basalts from Iceland and the Faeroes. The narrow range of initial 143Nd/144Nd ratios from 0.513026 ± 24 to 0.51292 ± 19 overlaps with the 143Nd/144Nd ranges for the other data fields shown, as well as with those for continental plateau basalts of Tertiary age from East and West Greenland (Carter et al. 1979).

In contrast, the initial 87Sr/86Sr ratios range from values typical for Atlantic oceanic basalts (ca. 0.703) up to higher values at ca. 0.705. The higher 87Sr/86Sr ratios result in a trend to the right of the anticorrelated array of 143Nd/144Nd and 87Sr/86Sr ratios (Fig. 1) documented for unaltered oceanic ridge basalts and some oceanic island basalts (e.g., O’Nions et al., 1977; Richard et al., 1976; De Paolo and Wasserburg, 1976). The effects of the acid leaching procedure on the 87Sr/86Sr ratios are indicated in Table 3 and Figure 1. The acid leached samples show a reduction both in the 87Sr/86Sr ratio and in the range of 87Sr/86Sr compared with the untreated samples. This results in Figure 1 in a data array more closely conformable with the main oceanic mantle array, although the two deepest samples still retain 87Sr/86Sr ratios that are relatively high for their respective 143Nd/144Nd ratios.

INTERPRETATION

Age of the Basalts and Its Significance

The argon analyses listed in Table 2 show a considerable decrease in atmospheric argon contamination as compared to those in Table 1. As a result the calculated ages are more precise. Nonetheless they remain mark-
Table 2. Second series of K-Ar analyses of samples A–G.

<table>
<thead>
<tr>
<th>Sample</th>
<th>Mass (g)</th>
<th>$^{40}$Ar$^{36}$Ar (× 10$^{-11}$ moles/g)</th>
<th>$^{40}$Ar$^{36}$Ar (× 10$^{-12}$ moles/g)</th>
<th>“Age” (m.y.)</th>
</tr>
</thead>
<tbody>
<tr>
<td>A</td>
<td>2.86897</td>
<td>1.810</td>
<td>4.222</td>
<td>66.3 ± 3.1</td>
</tr>
<tr>
<td>B</td>
<td>4.12175</td>
<td>6.916</td>
<td>16.78</td>
<td>199 ± 9</td>
</tr>
<tr>
<td>C</td>
<td>2.92494</td>
<td>2.206</td>
<td>4.971</td>
<td>52.3 ± 1.7</td>
</tr>
<tr>
<td>D</td>
<td>4.47448</td>
<td>3.237</td>
<td>4.899</td>
<td>54.5 ± 2.0</td>
</tr>
<tr>
<td>E</td>
<td>4.83465</td>
<td>34.83</td>
<td>30.4</td>
<td>397 ± 16</td>
</tr>
<tr>
<td>F</td>
<td>3.82658</td>
<td>4.771</td>
<td>12.54</td>
<td>110 ± 5</td>
</tr>
</tbody>
</table>

Note: Legend as for Table 1. For interpretation of results see text. Mean age of Samples C and D is 53.4 ± 1.3 m.y.

Table 3. Strontium and neodymium isotopic analyses of DSDP samples.

<table>
<thead>
<tr>
<th>Sample</th>
<th>$^{87}$Rb</th>
<th>$^{87}$Sr</th>
<th>$^{86}$Sr</th>
<th>$^{87}$Sr/$^{86}$Sr</th>
<th>$^{87}$Sr/$^{86}$Sr L</th>
<th>$^{87}$Sr/$^{86}$Sr R</th>
<th>$^{147}$Sm/$^{144}$Nd</th>
<th>$^{143}$Nd/$^{144}$Nd</th>
</tr>
</thead>
<tbody>
<tr>
<td>A</td>
<td>0.660</td>
<td>66.3</td>
<td>0.0288</td>
<td>0.70297(4)</td>
<td>0.70306(3)</td>
<td>0.70264(4)</td>
<td>5.70</td>
<td>0.329</td>
</tr>
<tr>
<td>B</td>
<td>0.723</td>
<td>74.3</td>
<td>0.0282</td>
<td>0.70319(4)</td>
<td>0.70301(4)</td>
<td>0.70266(4)</td>
<td>5.25</td>
<td>0.299</td>
</tr>
<tr>
<td>C</td>
<td>0.233</td>
<td>102.0</td>
<td>0.0148</td>
<td>0.70459(4)</td>
<td>0.70277(5)</td>
<td>0.70306(5)</td>
<td>10.6</td>
<td>0.218</td>
</tr>
<tr>
<td>D</td>
<td>0.612</td>
<td>105.0</td>
<td>0.0166</td>
<td>0.70329(7)</td>
<td>0.70406(4)</td>
<td>0.70336(4)</td>
<td>3.87</td>
<td>0.234</td>
</tr>
<tr>
<td>E</td>
<td>0.603</td>
<td>105.0</td>
<td>0.0201</td>
<td>0.70498(4)</td>
<td>0.70582(4)</td>
<td>0.70364(4)</td>
<td>1.27</td>
<td>0.227</td>
</tr>
<tr>
<td>F</td>
<td>1.05</td>
<td>96.1</td>
<td>0.0317</td>
<td>0.70522(4)</td>
<td>0.70633(3)</td>
<td>0.70396(3)</td>
<td>3.59</td>
<td>0.247</td>
</tr>
<tr>
<td>G</td>
<td>1.62</td>
<td>74.8</td>
<td>0.0627</td>
<td>0.70504(3)</td>
<td>0.70433(6)</td>
<td>0.70384(4)</td>
<td>4.15</td>
<td>0.319</td>
</tr>
</tbody>
</table>

$^a$ Concentrations determined by isotope dilution are in ppm and are each determined with a precision of ca. 0.5%.
$^b$ Initial ratios calculated for a crystallization age of 53 m.y.
$^c$ $^{143}$Nd/$^{144}$Nd for BCR in this laboratory = 0.512633 ± 10 (2σ, N = 4).
It has already been suggested that this episode also included the North Atlantic tuffs (main "ash marker") and East Greenland volcanism (Macintyre et al., 1979). Both of these are now known on biostratigraphic evidence to have occurred at the Paleocene/Eocene boundary (Smythe et al., 1983; Soper, Downie et al., 1976; Soper, Higgins, et al., 1976). It has further been demonstrated that the "ash marker" passes laterally into the Faeroese lavas (Smythe et al., 1983). This extensive igneous activity immediately precedes Anomaly 24B (and place this boundary at 54.9 m.y.). The samples analyzed in this study also belong to this zone. If, as seems probable, they also (like the tuffs and lavas of East Greenland and the Faeroes) form part of this far-reaching paroxysmal igneous event, which heralded the breakup of the North Atlantic, then the age of 53.5 ± 1.9 m.y. reported here can, in addition, be considered a measurement of the position of the Palaeocene/Eocene boundary on the geological timescale. This age may be compared with the assignment of ca. 54 m.y. on the 1964 Geological Society Phanerozoic timescale and the 51 m.y. date recently assigned by Odin (1982).

At 50 m.y. ago, seawater would have had a \(^{87}\text{Sr}/^{86}\text{Sr}\) ratio of ca. 0.708 (Hart and Staudigel, 1978; Hawkesworth and Elderfield, 1978) and its interaction with basaltic crust during hydrothermal alteration would result in raised and more variable \(^{87}\text{Sr}/^{86}\text{Sr}\) in the crust. The \(^{143}\text{Nd}/^{144}\text{Nd}\) composition of the basaltic crust would be little affected by such alteration because of the low abundance of Nd (ca. 3 ng \(1^{-1}\)) in seawater (e.g., Pieper and Wasserburg, 1980). This results in a trend in the right on the \(^{143}\text{Nd}/^{144}\text{Nd}\) vs. \(^{87}\text{Sr}/^{86}\text{Sr}\) diagram as indicated and as observed for the DSDP Leg 81 data (Fig. 1). The effect of such perturbation of pristine \(^{87}\text{Sr}/^{86}\text{Sr}\) may be largely removed by acid leaching as demonstrated by O’Nions and Pankhurst (1976) for altered DSDP Leg 37 basalts and as is evident for some of DSDP Leg 81 samples (Table 3, Fig. 1). The three deepest samples (E, F, G), however, retain relatively high \(^{87}\text{Sr}/^{86}\text{Sr}\) after acid leaching. This suggests retention of a marine Sr isotopic signature in an acid resistant phase(s), which may have formed at the higher temperatures of alteration experienced at these deeper levels (see Harrison et al., this volume).

**Nature and Origin of the Extraneous Argon**

Data from Tables 1 and 2 are displayed in Figure 2 in the form of a plot of total argon-40 vs. total argon-36. This graphically illustrates the strong correlation between these parameters referred to previously.

Samples A, C, and D plot toward the origin and scatter about line II with slope 286 ± 5 (details of all regression analyses will be found in Appendix C). This is similar to the atmospheric argon-40/36 ratio. The apparent collinearity results from the approximate equivalences in age and potassium content among these samples and the intercept corresponds roughly with their similar radiogenic argon concentrations (cf. Table 2).

Samples B, E, F, and G also have similar potassium contents and are presumed to be of similar age. They have, however, much higher argon concentrations, here attributed to the presence of extraneous argon. The data from Table 1 plot near line I with slope 304 ± 2 which lies slightly above line AR of atmospheric slope. When the relative magnitude of the atmospheric argon contamination is reduced (data from Table 2) the points then define another line III with slope 436 ± 2 and negative intercept on the ordinate.

A rudimentary model is proposed to explain these results. This simply envisages that some of the samples, which are of the same age and similar potassium contents, have been contaminated, to a greater or lesser degree, by the incorporation of varying amounts of extraneous argon of fixed isotopic composition. A bulk contamination model of this type is considered to be a

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The isotopic ratio appears to be quite uniform over a considerable depth range, but there is a marked increase in argon concentration with depth followed by an apparent decrease (Table 2). This seems to reflect the variation in initial strontium isotopic composition \((^{87}Sr/^{86}Sr)_{i}\) in Table 3. The extraneous argon may therefore also have been derived through interaction with hot fluids (for which there is ample mineralogical and isotopic evidence) circulating in the pile during burial. The source of the argon was presumably earlier-formed K-bearing detrital material in sediments, which degassed as these were dewatered during diagenesis.

The absence of extraneous argon in the rocks from higher levels suggests that the interactions described above did not accompany eruption but occurred during burial soon afterwards. The extraneous argon recognized in these rocks is neither exclusively excess or inherited in the classification of Dalrymple and Lanphere (1969). It is in a sense both.

**CONCLUSIONS**

Direct measurement of the K-Ar age of samples from Sites 553 and 555 is complicated by the presence of an extraneous argon component whose concentration increases with depth. Nevertheless an age of 53.5 \(\pm\) 1.9 m.y. can be determined for the uppermost samples at Site 555. The extraneous component appears to have an argon-40/argon-36 ratio of \(\geq 436\) and to have been derived from sediments and incorporated in the samples from fluids circulating during burial.

The measured age is similar to that of extensive igneous activity around the margins of the North Atlantic. However, it is of additional significance in that it can be regarded as a good estimate of the age of Anomaly 24B on the geomagnetic timescale, and possibly also of the Paleocene/Eocene boundary on the geological timescale. To the best of our knowledge, this is the first direct measurement of the age of a marine magnetic anomaly.

Apart from a marine strontium isotopic signature the results of the trace element and isotopic analyses are very similar to those which have previously been derived for other Atlantic Ocean basalts. No evidence has been found, in any of the rocks analyzed, for any significant contribution of the continental crust to their genesis.

It must again be emphasized that little evidence for crustal contamination has been found from the Sr and Nd data from some plateau basalts which are demonstrably continental, such as those in East and West Greenland and the Faeroes. In the latter, three silicic basalts with high strontium initial ratios, which are probably the result of continental crustal contamination, have only recently been identified (Haald and Waagstein, 1983; see also caption, Fig. 2). The Pb isotopic data are, however, indicative of crustal contamination (Gariépy et al., 1983). The Sr and Nd isotopic results are therefore equivocal in that they do not necessarily imply the absence of continental crust (or extreme thinning, rapid magma ascent, etc.), although models involving only oceanic crust (e.g., Smythe et al., 1983) perhaps are speculatively favored. We therefore conclude by reiterating (paraphrasing Bradley, 1962) that “there is confusion in any
reasoning which regards the absence of evidence for continental crust as equivalent to evidence for the absence of continental crust."

ACKNOWLEDGMENTS

We thank Mr. J. Gray, Mr. J. Jocelyn and Mr. J. Hutchison for technical assistance, Ms. J. Bennett for drafting the figures and Ms. J. McEwan for typing. Financial support from the Scottish Universities and N.E.R.C. is gratefully acknowledged.

REFERENCES


APPENDIX A

Identification Numbers of Samples

A = Sample 553-46-2, 31–33 cm; F8480
B = Sample 553-49-4, 93–96 cm; F8484
C = Sample 555-69-2, 21–24 cm; glomeroporphyritic basalt: FS497
D = Sample 555-69-4, 85–88 cm; phric basalt: F8498
E = Sample 555-76-1, 69–74 cm; dolerite: F8501
F = Sample 555-90-1, 100–104 cm; aphric basalt: F8504
G = Sample 555-96-3, 110–113 cm; olivine dolerite: F8508

APPENDIX B

Explanation of Figure 2

As explained in the text, we confine ourselves specifically here to the special case in which the radiogenic argon–40 concentrations and isotopic composition of the extraneous argon are assumed to be approximately constant among a series of samples.

Let the subsets T, E, R, and A refer to the total, extraneous, radiogenic, and atmospheric concentrations, respectively. Then in any analysis we have:

\[ 40_\text{T} = 40_\text{E} + 40_\text{R} + 40_\text{A} \]

and

\[ 36_\text{T} = 36_\text{E} + 36_\text{A} \]

Also let

\[ R_\text{E} = \frac{40_\text{E}}{36_\text{E}} \]

and\[ R_\text{A} = \frac{40_\text{A}}{36_\text{A}} \]

(= 295.5)

then

\[ 40_\text{T} = 36_\text{T} R_\text{E} - \frac{R_\text{E} - R_\text{A}}{R_\text{A}} (40_\text{E} + 40_\text{R}) \]

i.e.,

\[ 40_\text{T} = 36_\text{T} R_\text{E} - \frac{R_\text{E} - R_\text{A}}{R_\text{A}} (40_\text{E} + 40_\text{R}) \]

In Figure 2, we plot 40_\text{T} vs. 36_\text{T}. We first recognize the two extreme cases, that is:

1. If R_\text{E} = R_\text{A}, the points will lie on straight line with slope = R_\text{A} and intercept 40_\text{E} on 40_\text{T} - axis (point "R" on Fig. 2).
2. If 40_\text{A} = 0 the points are again collinear and the line passes through "R", but this time the slope is R_\text{A}.

In intermediate cases, where neither of these conditions is satisfied, a straight line will again result as long as the second term on the RHS is essentially constant, i.e., if 40_\text{A} is constant. If additionally this term is small, in comparison to 40_\text{E} the slope of the line will be a good approximation to the ratio, R_\text{A}. Unless R_\text{E} is exceptionally large, this latter assumption is equivalent to assuming that 40_\text{A} is small. Thus, in these intermediate cases, the intercept is less than the radiogenic point by an amount equal to 40_\text{A} (R_\text{E} - R_\text{A} - 1). The magnitude of this deviation (as compared to 40_\text{E}) is therefore a measure of the closeness of approach of the slope to the ratio R_\text{E}.
For a particular gas-handling system, 40A can usually be quite reliably estimated from general laboratory experience. For a constant line-blank it will be much smaller when the sample mass is increased. The best estimate can sometimes, as in this case, be derived from the analyses of other samples in the suite which contain no extraneous argon. Thus for samples C and D it is readily apparent from Figure 2 and from Table 2 that 40A (= 40R - 40b) = 0.8 x 10^-11 moles/g and that this is small in comparison to most measured values of 40R (≥ 3.8 x 10^-11 moles/g).

The slope of the line, which is in any case determined primarily by the data points with high 40R, can therefore in this case be considered to be a good approximation to the isotopic composition of the extraneous argon (i.e., R_E ≥ 436). This permits evaluation of the extraneous argon-36 concentrations (36E) since the argon-40 concentrations (40E) are known (= 40R - 40b - 40A).

Alternative approach. In the previous construct, as R_E increases so also does the slope and the departure of the intercept from the radiogenic point. The presence of a (constant) 40A means that the measured slope < R_E and the closeness of approach is governed by the relative magnitude of the atmospheric contamination. This is least for points with high 40R and 36R. If we take one such point (e.g., E on Fig. 2) and constrain a line of slope m through this point, such that its intercept differs from the radiogenic point by an amount 40A (m/R_A - l), then m should be a very good approximation to R_E.

It is a relatively simple matter to show that this condition is fulfilled if m satisfies (1 - f_36)m = m_R - l, where m_R is the slope of the join to the radiogenic point, R, f_36 = 36A/36R, and l = 40A/36R.

As before, for small values of f, m = m_R. Usually f_36 << m_R, and so for small values of f, the slope which best corresponds to the ratio R_E is given by m = m_R/1 - f_36. This demonstrates the critical dependence of m on the magnitude of the atmospheric argon-36 contamination.

Considering analysis E in Figure 2 and Table 2 and again assuming that 40A = 8 x 10^-12 moles/g, then f_36 ~ 0.076 and m_R = 409. Thus R_E = m = 443, which is close to the value derived previously.

It is perhaps also worth noting that the above conclusions regarding the isotopic composition of the extraneous argon require no a priori knowledge of the age or potassium content of the samples (only an approximate equivalence of their radiogenic argon-40 concentrations). Thus if the same, albeit rather unique, circumstances can be assumed, this simple graphical approach might prove useful when these parameters are unknown or difficult to measure, e.g., in a suite of contemporaneous oceanic basaltic glasses.

### APPENDIX C

#### Details of Regression Analyses

<table>
<thead>
<tr>
<th>Line</th>
<th>No. of points</th>
<th>Error X (%)</th>
<th>Error Y (%)</th>
<th>Slope and error</th>
<th>Intercept and error</th>
<th>MSWD</th>
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</thead>
<tbody>
<tr>
<td>I</td>
<td>4</td>
<td>1</td>
<td>0.5</td>
<td>304 ± 2</td>
<td>0.50 ± 0.06</td>
<td>0.09</td>
</tr>
<tr>
<td>II</td>
<td>8</td>
<td>1</td>
<td>0.5</td>
<td>286 ± 5</td>
<td>0.52 ± 0.02</td>
<td>2.46</td>
</tr>
<tr>
<td>I + II</td>
<td>12</td>
<td>1</td>
<td>0.5</td>
<td>300 ± 3</td>
<td>0.47 ± 0.02</td>
<td>4.58</td>
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<tr>
<td>III</td>
<td>4</td>
<td>1</td>
<td>0.5</td>
<td>436 ± 2</td>
<td>-0.44 ± 0.03</td>
<td>0.12</td>
</tr>
<tr>
<td>IV</td>
<td>5</td>
<td>2</td>
<td>0.5</td>
<td>59.0 ± 3.1</td>
<td>295 ± 3</td>
<td>2.90</td>
</tr>
<tr>
<td>V</td>
<td>7</td>
<td>2</td>
<td>0.5</td>
<td>53.5 ± 1.9</td>
<td>295 ± 3</td>
<td>2.79</td>
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</tbody>
</table>

Note: All analyses carried out following method of York (1969). X (%) and Y (%) are blanket errors assigned to coordinates. Errors on ages and intercepts are scatter errors (1σ). For IV and V slope is presented as age in m.y.